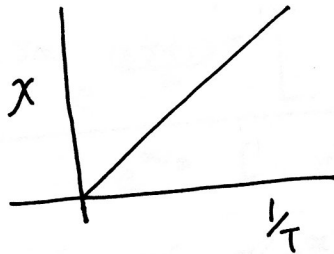
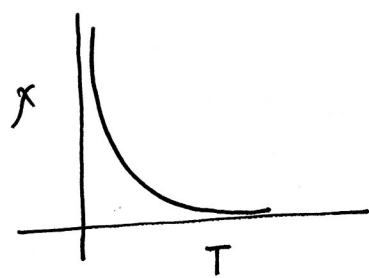


$$\chi = \frac{C}{T}$$

$$\text{where } C = \frac{n \mu_0 \mu_B^2}{k_B}$$

Curie's Law

Temperature dependence of susceptibility,



Now,

We consider the case for arbitrary angular momentum J ,
There are $m_J = -J, -J+1, \dots, J-1, +J$ values (i.e. J is quantized)

$$\mu_z = g_J \mu_B m_J$$

$$E = -\vec{\mu} \cdot \vec{B} = -\mu_z B = -g_J \mu_B m_J B \quad (1)$$

(Energy splits into $(2J+1)$ states)

$$M = \frac{N}{V} \frac{\sum_{m_J} (g_J \mu_B m_J) e^{\frac{g_J \mu_B m_J B}{k_B T}}}{Z} = \frac{N}{V} \frac{\sum_{m_J} (g_J \mu_B m_J) e^{m_J \alpha}}{Z}, \quad \alpha = \frac{g_J \mu_B B}{k_B T} \quad (2)$$

The partition function,

$$Z = \sum_{m_J} e^{\frac{g_J \mu_B m_J B}{k_B T}} = \sum_{m_J} e^{m_J \alpha} \quad (3)$$

$$\frac{\partial Z}{\partial \alpha} = \sum_{m_J} m_J e^{m_J \alpha} \quad (4)$$

$$M = \frac{N}{V} g_J \mu_B \sum_{m_J} \frac{m_J e^{m_J \alpha}}{Z}$$

$$= \frac{N}{V} g_J \mu_B \frac{1}{Z} \frac{\partial Z}{\partial \alpha}$$

$$M = \frac{N}{V} g_J \mu_B \frac{\partial (\ln Z)}{\partial \alpha} \quad (5)$$

from eqⁿ (3),

$$\begin{aligned}
 Z &= \sum_{m_J} e^{m_J x} = \sum_{m_J=-J}^J (e^x)^{m_J} \quad \left[\begin{array}{l} \text{Sum of Geometric} \\ \text{series} \\ S_n = \frac{a_0(1-r^n)}{1-r} \end{array} \right] \\
 &= \frac{e^{-Jx} [1 - e^{(2J+1)x}]}{1 - e^x} \\
 &= \frac{e^{-Jx} e^{\frac{(2J+1)x}{2}} \left[e^{-\frac{(2J+1)x}{2}} - e^{\frac{(2J+1)x}{2}} \right]}{e^{x/2} [e^{-x/2} - e^{x/2}]} \\
 &= e^{-Jx + Jx + \frac{x}{2} - \frac{x}{2}} \frac{\left[e^{-\frac{(2J+1)x}{2}} - e^{\frac{(2J+1)x}{2}} \right]}{[e^{-x/2} - e^{x/2}]}
 \end{aligned}$$

$$Z = \frac{\sinh \left[\left(J + \frac{1}{2} \right) x \right]}{\sinh \left(\frac{x}{2} \right)} \quad \text{--- (6)}$$

~~Substituting~~

Taking log of Z,

$$\ln Z = \ln \sinh \left[\left(J + \frac{1}{2} \right) x \right] - \ln \sinh \left(\frac{x}{2} \right)$$

$$\frac{\partial (\ln Z)}{\partial x} = \frac{\cosh \left[\left(J + \frac{1}{2} \right) x \right]}{\sinh \left[\left(J + \frac{1}{2} \right) x \right]} \cdot \left(\frac{2J+1}{2} \right) - \frac{\cosh \left(\frac{x}{2} \right)}{\sinh \left(\frac{x}{2} \right)} \cdot \left(\frac{1}{2} \right) \quad \text{--- (7)}$$

substituting (7) in (5)

$$M = \left(\frac{N}{V} g_J \mu_B \right) J \left[\left(\frac{2J+1}{2J} \right) \coth \left(J + \frac{1}{2} \right) x - \frac{1}{2J} \coth \left(\frac{x}{2} \right) \right]$$

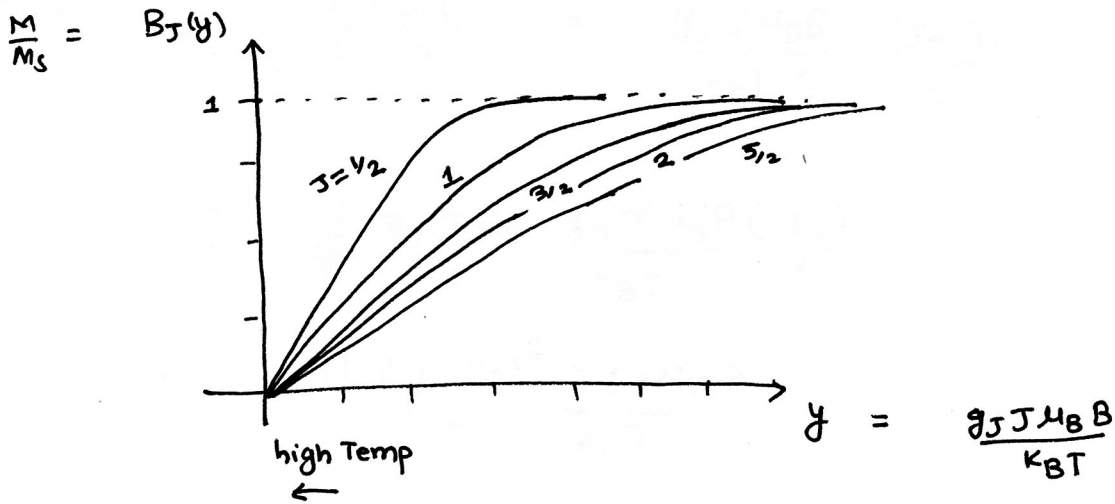
$$= \left(\frac{N}{V} g_J J \mu_B \right) \left[\left(\frac{2J+1}{2J} \right) \coth \left(\frac{2J+1}{2J} \right) Jx - \frac{1}{2J} \coth \left(\frac{xJ}{2J} \right) \right]$$

$$M = M_s \left[\left(\frac{2J+1}{2J} \right) \coth \left(\frac{2J+1}{2J} \right) y - \frac{1}{2J} \coth \frac{1}{2J} y \right]$$

$$\begin{aligned}
 \text{where } y &= xJ \\
 &= \frac{g_J J \mu_B B}{k_B T}
 \end{aligned}$$

$$\frac{M}{M_s} = B_J(y) \quad , \text{ where } B_J(y) \text{ is the Brillouin function.}$$

Plot of the Brillouin function $B_J(y)$ for various values of the spin J .



(a) For fixed B , as $T \rightarrow 0$, $M = M_s$ i.e. each ion is completely aligned by the field. However this case arises only when $k_B T \ll g_J \mu_B B$ i.e. $T < 1$ K in a field. $T \ll \frac{g_J \mu_B B}{k_B} \approx 1$ K for field of 1 T.

(b) At high temperatures, $g_J J \mu_B B \ll k_B T$,

$$\coth x \approx \frac{1}{x} + \frac{1}{3}x + O(x^3),$$

$$B_J(y) \approx \left(\frac{2J+1}{2J}\right) \left\{ \frac{1}{\left(\frac{2J+1}{2J}\right)y} + \frac{1}{3} \left(\frac{2J+1}{2J}\right)y \right\} - \frac{1}{2J} \left\{ \frac{2J}{y} + \frac{1}{3} \frac{y}{2J} \right\}$$

$$\approx \frac{1}{y} + \frac{1}{3} \left(\frac{2J+1}{2J}\right)^2 y - \frac{1}{y} - \frac{1}{3} \frac{y}{(2J)^2}$$

$$\approx \frac{y}{3} \left[\left(\frac{2J+1}{2J}\right)^2 - \left(\frac{1}{2J}\right)^2 \right] \approx \frac{y}{3} \left[\left(\frac{2J+1}{2J} + \frac{1}{2J}\right) \left(\frac{2J+1-1}{2J}\right) \right]$$

$$B_J(y) \approx \frac{y}{3} \frac{(J+1)}{J}$$

$$\frac{M}{M_s} = \frac{g}{3} \frac{J(J+1)}{J} = \frac{g_J J \mu_B B}{k_B T} \frac{J(J+1)}{J}$$

$$M = \frac{N}{V} g_J J \mu_B \frac{g_J J \mu_B B (J+1)}{k_B T}$$

$$M = \frac{N}{V} \frac{(g_J \mu_B)^2}{3} \frac{J(J+1) B}{k_B T}$$

$$\chi = \frac{M}{H} = \frac{N}{V} \frac{(g_J \mu_B)^2 \mu_0}{3} \frac{J(J+1)}{k_B T}$$

Susceptibilities are usually quoted as molar susceptibilities, based on the magnetization per mole, rather than per cubic centimeter. Thus,

χ_{molar} is given by multiplying χ by the volume of a mole, $\frac{N_A}{[N/V]}$

$$\chi_{\text{molar}} = \frac{N_A (g_J \mu_B)^2 \mu_0}{3 k_B T} J(J+1)$$

; where $N_A = \text{Avogadro's no}$

Thus, the susceptibility is inversely proportional to temperature.

This variation of χ with T is ~~called~~ known as Curie's Law.

~~The~~ The effective Bohr magneton ~~number~~ μ_{eff} is given by,

$$\mu_{\text{eff}} = g_J \sqrt{J(J+1)} \mu_B$$

$$\Rightarrow \mu_{\text{eff}}^2 = g_J^2 J(J+1) \mu_B^2$$

$$\therefore g_J = \frac{3}{2} + \frac{1}{2} \left[\frac{S(S+1) - L(L+1)}{J(J+1)} \right]$$

The Curie's Law can be written as,

$$\chi_{\text{molar}} = \frac{N_A \mu_0 \mu_{\text{eff}}^2}{3 k_B T}$$