

what is the spin-Hamiltonian (Heisenberg)

1) It has to be such that $H|\psi_s\rangle = E_s|\psi_s\rangle$
 $H|\psi_t\rangle = E_t|\psi_t\rangle$

H is the 2-electron Hamiltonian we treated before
 $E_s \neq E_t$ because of Coulomb repulsion (e^2/r_{12})

$\psi_s \rightarrow$ singlet state with eigenvalue E_s , $S=0$

$\psi_t \rightarrow$ triplet state with E_t , $S=1$

~~$S = 1$~~ $S^2 = (S_1 + S_2)^2 = S_1^2 + S_2^2 + 2S_1 \cdot S_2$

$$S_1^2 = 1/2(1/2+1) = S_2^2 = 3/4$$

For $S=0 \Rightarrow 0 = 2 \cdot 3/4 + 2S_1 \cdot S_2 \Rightarrow S_1 \cdot S_2 = -3/4$

For $S=1 \Rightarrow 2 = 2 \cdot 3/4 + 2S_1 \cdot S_2 \Rightarrow S_1 \cdot S_2 = +1/4$

So eigenvalue of $S_1 \cdot S_2 = -3/4$ for singlet state
 $= +1/4$ for triplet state.

~~with~~ Construct a Hamiltonian

$$H = \frac{1}{4}(E_s + 3E_t) - (E_s - E_t)S_1 \cdot S_2$$

what is $\langle \psi_s | H | \psi_s \rangle = \frac{1}{4}(E_s + 3E_t) \langle \psi_s | \psi_s \rangle - (E_s - E_t) \langle \psi_s | S_1 \cdot S_2 | \psi_s \rangle$
 $= \frac{1}{4}(E_s + 3E_t) - \frac{3}{4}(E_s - E_t) = E_s$

What is $\langle \psi_t | H | \psi_t \rangle = \frac{1}{4}(E_s + 3E_t) - \frac{1}{4}(E_s - E_t) = E_t$

So H is the spin-Hamiltonian.

Ignoring the constant

$$H_{\text{spin}} = -(E_s - E_t) S_1 \cdot S_2$$

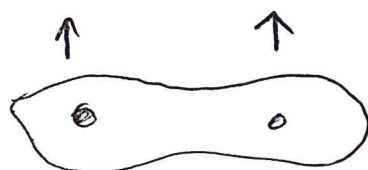
$$\boxed{H_{\text{spin}} = -J S_1 \cdot S_2} \quad J = (E_s - E_t)$$

When $E_s - E_t > 0$ S_1 is parallel to S_2 (FM) is lower in energy.

When $E_s - E_t < 0$ S_1 is anti parallel to S_2 (AFM)

Whether $E_s > E_t$ or $E_t > E_s$ depends on the spin-independent Hamiltonian (orbital/space) through Pauli exchange and Coulomb interaction

In the case of H_2/O_2 molecule $E_s < E_t$



$S=1$
not low energy

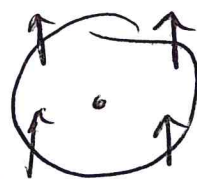
electrons in the middle



$S=0$
low energy

↳ gain energy through nuclear attraction from both ~~electron~~ nuclei (singlet state)

But in a single atom



low energy ($S=1$) Hund's rule

Pauli exclusion does not allow electrons to come close which minimizes Coulomb exchange