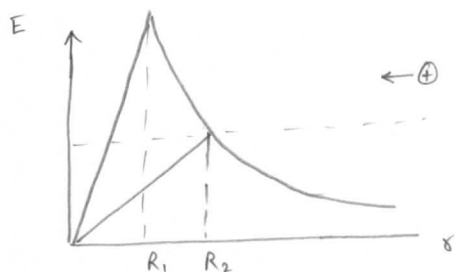


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① Rutherford model of atom

$$\vec{E} = \begin{cases} \frac{kQ}{R^2} \frac{r}{R} \hat{r} & r < R \\ \frac{kQ}{r^2} \hat{r} & r > R \end{cases}$$



② Electric field of a hydrogen atom.
Quantum mechanically the probability density of finding an electron around the proton is

$$P(r) = \frac{1}{8\pi} \left(\frac{2}{a_0}\right)^3 e^{-\frac{2r}{a_0}}$$

where a_0 is the Bohr radius.

$$a_0 = \frac{\hbar}{m_e c \alpha} \sim 5 \times 10^{-11} \text{ m}$$

Note that

$$\begin{aligned} \int d^3r P(r) &= 4\pi \frac{1}{8\pi} \left(\frac{2}{a_0}\right)^3 \int_0^\infty r^2 dr e^{-\frac{2r}{a_0}} \\ &= \frac{1}{2} \int_0^\infty \frac{dt}{t} t^3 e^{-t} = 1 \quad \checkmark \end{aligned}$$

- ③ Let us calculate the electric field in and around the hydrogen atom. (Assume proton is a point particle.)



$$\rho(r) = -e P(r)$$

↳ for electron

$$E 4\pi R^2 = \frac{Q_{in}}{\epsilon_0}$$

$$E = \frac{1}{4\pi\epsilon_0} \frac{1}{R^2} Q_{in}$$

$$\begin{aligned} Q_{in} &= +e - e \int_R d^3r P(r) \\ &= e \left[1 - 4\pi \int_0^R r^2 dr \frac{1}{8\pi} \left(\frac{2}{a_0}\right)^3 e^{-\frac{2r}{a_0}} \right] \\ &= e \left[1 - \frac{1}{2} \int_0^{\frac{2R}{a_0}} \frac{dt}{t} t^3 e^{-t} \right] \\ &= e \frac{1}{2} \int_{\frac{2R}{a_0}}^{\infty} \frac{dt}{t} t^3 e^{-t} \\ &= e e^{-\frac{2R}{a_0}} \left[1 + \frac{2R}{a_0} + \frac{1}{2} \left(\frac{2R}{a_0}\right)^2 \right] \end{aligned}$$

$$\begin{aligned} &\int_x^{\infty} dt t^2 e^{-t} \\ &= x^2 - \int_x^{\infty} dt 2t \frac{d}{dt} e^{-t} \\ &= x^2 e^{-x} + 2x e^{-x} + 2e^{-x} \\ &= e^{-x} [x^2 + 2x + 2] \end{aligned}$$

Thus,

$$E = \frac{1}{4\pi\epsilon_0} \frac{e}{R^2} \left[1 - \frac{1}{6} \left(\frac{2R}{a_0}\right)^3 + O\left(\frac{2R}{a_0}\right)^4 \right]$$

$$\frac{1}{4\pi\epsilon_0} \frac{e}{a_0^2} \left[1 + \frac{2R}{a_0} + \frac{1}{2} \left(\frac{2R}{a_0}\right)^2 \right] e^{-\frac{2R}{a_0}}$$

for $2R < a_0$

for $a_0 < 2R$

(4)

ExampleSolid sphere with $\rho(r) = A r^n$

$$Q = \int d^3r \rho(r) = 4\pi \int_0^R r^2 dr A r^n = 4\pi A \frac{R^{n+3}}{(n+3)}$$

$$A = \frac{(n+3) Q}{4\pi R^{n+3}}$$

Using Gauss's law

$$E 4\pi r^2 = \frac{Q_{in}}{\epsilon_0}$$

$$E = \frac{1}{4\pi\epsilon_0} \frac{1}{r^2} Q_{in}$$

for $r < R$:

$$\begin{aligned} Q_{in} &= 4\pi \int_0^r r'^2 dr' A r'^n \\ &= 4\pi A \frac{r^{n+3}}{n+3} \\ &= 4\pi \frac{(n+3) Q}{4\pi R^{n+3}} \frac{r^{n+3}}{n+3} \\ &= Q \left(\frac{r}{R}\right)^{n+3} \end{aligned}$$

Thus,

$$E = \begin{cases} \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2} \left(\frac{r}{R}\right)^{n+3}, & r < R, \\ \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}, & R < r. \end{cases}$$

This leads to the earlier evaluated expression
for $n=0$ (uniform charge distribution).