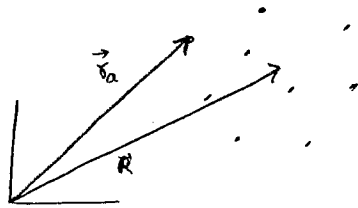


Date: 2014 Oct 20

① An atom consists of charges (electrons and protons)



$\vec{R}$ - arbitrary point, say center of charge.

$$q_{\text{atom}} = \sum_a q_a = 0$$

$$\vec{d}_{\text{atom}} = \sum_a q_a (\vec{r}_a - \vec{R})$$

$$= \sum_a q_a \vec{r}_a - \vec{R} \underbrace{\sum_a q_a}_{=0}$$

$$= \sum_a q_a \vec{r}_a$$

② Thus, an atom in the presence of an electric field interacts with the electric field due to its dipole moment. The interaction energy, and the corresponding force and torque, for a dipole interacting with a electric field is

$$U = - \vec{d} \cdot \vec{E}$$

$$\vec{F} = - \vec{\nabla} U = \vec{\nabla} (\vec{d} \cdot \vec{E}) = (\vec{d} \cdot \vec{\nabla}) \vec{E}$$

$$\vec{\tau} = \vec{d} \times \vec{E}$$

③ Using ① and ② together we have the force and torque on an individual atom to be

$$\vec{F}_{\text{atom}} = \vec{d}_{\text{atom}} \cdot \vec{\nabla} \vec{E}(\vec{R})$$

$$\vec{\tau}_{\text{atom}} = \vec{d}_{\text{atom}} \times \vec{E}(\vec{R})$$

④ In particular, from scratch again,

$$\begin{aligned} \vec{F}_{\text{atom}} &= \sum_a q_a \vec{E}(\vec{r}_a) \\ &= \underbrace{\sum_a q_a}_{=0} \vec{E}(\vec{R}) + \underbrace{\sum_a q_a (\vec{r}_a - \vec{R})}_{\vec{d}_{\text{atom}}} \cdot \vec{\nabla} \vec{E}(\vec{R}) + \dots \\ &= \vec{d}_{\text{atom}} \cdot \vec{\nabla} \vec{E}(\vec{R}) \end{aligned}$$

⑤ Force on a macroscopic material is obtained by adding (or integrating) over all the individual atoms.

$$\vec{F} = \int d^3r \, n(\vec{r}) \vec{F}_{\text{atom}}$$

$$\begin{aligned} n(\vec{r}) &= \frac{\text{No. of atoms}}{\text{volume}} \\ &= \text{number density} \end{aligned}$$

$$= \int d^3r \, n(\vec{r}) \vec{d}_{\text{atom}} \cdot \vec{\nabla} \vec{E}$$

$$= \int d^3r \, \vec{P}(\vec{r}, t) \cdot \vec{\nabla} \vec{E}$$

$\vec{d}(\vec{r}, t) \rightarrow$ distribution of atomic dipole moments, which could have time dependence.

where

$$\vec{P}(\vec{r}, t) = n(\vec{r}) \vec{d}(\vec{r}, t)$$

$\vec{P}(\vec{r}, t) \rightarrow$ polarization of the material.

⑥ Thus, the force on a macroscopic material is given by

$$\begin{aligned}\vec{F} &= \int d^3r \vec{P}(\vec{r}, t) \cdot \vec{\nabla} \vec{E} \\ &= \int d^3r (-\vec{\nabla} \cdot \vec{P}) \vec{E} + \underbrace{\vec{P} \cdot \vec{E}}_{=0} \Big|_{\text{surface}} \\ &= \int d^3r (-\vec{\nabla} \cdot \vec{P}) \vec{E}\end{aligned}$$

because $n(\vec{r}) = 0$ on the surface.

⑦ Comparing this with

$$\vec{F} = q \vec{E} = \int d^3r \rho(\vec{r}, t) \vec{E}(\vec{r}, t)$$

we identify the effective charge density of a macroscopic material to be

$$\rho_{\text{eff}}(\vec{r}, t) = -\vec{\nabla} \cdot \vec{P}(\vec{r}, t)$$

⑧ Similarly, we can show that the effective current density of a macroscopic material is given by

$$\vec{j}_{\text{eff}}(\vec{r}, t) = \frac{\partial}{\partial t} \vec{P} + \vec{\nabla} \times \vec{M}$$

→ Magnetization of the material

$$\vec{M}(\vec{r}, t) = n(\vec{r}) \vec{\mu}(\vec{r}, t)$$

$$\vec{\mu} = \frac{1}{2} q_a \vec{r}_a \times \vec{v}_a$$

⑨ The Maxwell's equations in the presence of macroscopic materials reads.

$$\vec{\nabla} \cdot \epsilon_0 \vec{E} = \rho - \vec{\nabla} \cdot \vec{P}$$

$$\vec{\nabla} \times \vec{E} = -\frac{\partial \vec{B}}{\partial t}$$

$$\vec{\nabla} \times \frac{\vec{B}}{\mu_0} = \frac{\partial}{\partial t} \epsilon_0 \vec{E} + \vec{J} + \frac{\partial \vec{P}}{\partial t} + \vec{\nabla} \times \vec{M}$$

$$\vec{\nabla} \cdot \vec{B} = 0$$

which can be rewritten as.

$$\vec{\nabla} \cdot (\epsilon_0 \vec{E} + \vec{P}) = \rho$$

$$\vec{\nabla} \times \left(\frac{\vec{B}}{\mu_0} - \vec{M} \right) = \frac{\partial}{\partial t} (\epsilon_0 \vec{E} + \vec{P}) + \vec{J}$$

⑩ We can define

$$\vec{D} = \epsilon_0 \vec{E} + \vec{P}$$

$$\vec{H} = \frac{1}{\mu_0} \vec{B} - \vec{M}$$

to write the macroscopic Maxwell's equation

$$\vec{\nabla} \times \vec{E} = -\frac{\partial \vec{B}}{\partial t}$$

$$\vec{\nabla} \cdot \vec{D} = \rho$$

$$\vec{\nabla} \cdot \vec{B} = 0$$

$$\vec{\nabla} \times \vec{H} = \frac{\partial \vec{D}}{\partial t} + \vec{J}$$