## Homework No. 13 (Spring 2018) PHYS 530A: Quantum Mechanics II

Due date: Thursday, 2018 May 3, 4.30pm

- 1. (20 points.) The components of the position and momentum operator, **r** and **p**, respectively, satisfy the commutation relations  $[r_i, p_j] = i\hbar \delta_{ij}$ . Verify the following:
  - (a)  $\mathbf{r} \times \mathbf{p} + \mathbf{p} \times \mathbf{r} = 0.$
  - (b)  $\mathbf{r} \cdot \mathbf{p} \mathbf{p} \cdot \mathbf{r} = 3i\hbar$ .
  - (c)  $(\mathbf{a} \cdot \mathbf{r})(\mathbf{b} \cdot \mathbf{p}) (\mathbf{b} \cdot \mathbf{p})(\mathbf{a} \cdot \mathbf{r}) = i\hbar(\mathbf{a} \cdot \mathbf{b})$ , where **a** and **b** and numerical.
  - (d)  $\mathbf{r} \times (\mathbf{r} \times \mathbf{p}) = \mathbf{r} \, \mathbf{p} \cdot \mathbf{r} \mathbf{p} r^2 + i\hbar \mathbf{r}.$
- 2. (20 points.) Using commutation relations between r, p, and L, verify the following:
  - (a)  $\mathbf{L} \times \mathbf{L} = i\hbar \mathbf{L}$ .
  - (b)  $\mathbf{p} \times \mathbf{L} + \mathbf{L} \times \mathbf{p} = 2i\hbar \mathbf{p}.$
  - (c)  $-\mathbf{L} \times \mathbf{p} \cdot \frac{\mathbf{r}}{r} = L^2 \frac{1}{r} = \frac{1}{r} L^2.$
- 3. (30 points.) Using commutation relations between r, p, and L, verify the relation

$$\mathbf{p} \times \mathbf{L} \cdot \mathbf{p} = 2i\hbar \, p^2. \tag{1}$$

Thus, verify that either of the three equalities for

$$\mathbf{M} = -\frac{1}{2} \left( \mathbf{p} \times \mathbf{L} - \mathbf{L} \times \mathbf{p} \right) = -\mathbf{p} \times \mathbf{L} + i\hbar \mathbf{p} = \mathbf{L} \times \mathbf{p} - i\hbar \mathbf{p}$$
(2)

leads to

$$M^2 = (L^2 + \hbar^2)p^2.$$
 (3)

This ensures that either of the following three expressions for the Axial vector

$$\mathbf{A} = \hat{\mathbf{r}} - \frac{1}{\mu Z e^2} \frac{1}{2} \left( \mathbf{p} \times \mathbf{L} - \mathbf{L} \times \mathbf{p} \right)$$
(4a)

$$=\hat{\mathbf{r}} - \frac{1}{\mu Z e^2} \mathbf{p} \times \mathbf{L} + \frac{i\hbar}{\mu Z e^2} \mathbf{p}$$
(4b)

$$= \hat{\mathbf{r}} + \frac{1}{\mu Z e^2} \mathbf{L} \times \mathbf{p} - \frac{i\hbar}{\mu Z e^2} \mathbf{p}$$
(4c)

leads to

$$A^2 = 1 + \frac{2(L^2 + \hbar^2)H}{\mu Z^2 e^4}.$$
(5)

After using the Bohr quantization condition  $L = n' \hbar$ , where n' = 0, 1, 2, ..., and presuming that the orbit is a circle that has eccentricity A = 0, Eq. (5) leads to the energy levels predicted by the Bohr model,

$$H = -\frac{\mu Z^2 e^4}{\hbar^2} \frac{1}{2n^2}, \qquad n = 1, 2, 3, \dots$$
(6)

Show that the (classical) analysis of hydrogen atom, (that does not accommodate the Heisenberg uncertainty relation contained in the commutation relations between  $\mathbf{r}$  and  $\mathbf{p}$ ,) in addition, admits the n = 0 energy state which permits orbits of vanishing radius.